

AD 685902

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107 Years of Line-Formula Notations (1861-1968)*

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Received May 21, 1968

Within seven years after the birth of structural chemistry in 1861, virtually all of the main ideas relating to line-formula conventions were devised and published in the leading chemical journals of a century ago. No basically new practices appeared for some 79 years. Then, within an identically brief period of just seven years (1947-1954), virtually all of the fundamental features of structure-delineating chemical notations appeared in the international chemical literature. The key characteristics of the old conventions and new systems are surveyed.

"The recent international interest in chemical notation has made it seem profitable and desirable to examine the historical records for a guiding background." This was the opening remark for a report on "The History of Chemical Notation" presented 18 years ago at the 118th Meeting of the American Chemical Society in Chicago.¹ That report in turn quoted an opening remark on "Chemical Notation and Nomenclature" by Samuel William Johnson that has an amusing echo today because his remark now is nearly 100 years old:²

"Beginners in Chemistry are liable to much confusion and embarrassment from the fact that there are now in use two

distinct systems of Chemical Notation and several forms of nomenclature."

Johnson was referring to notation developments that were then just 10 years old. More recent details on the "Origin of the Line-Formula Method" were given in 1954,³ and these historic developments again were reported to the American Chemical Society in 1962, this time with a century-old perspective.⁴

The key idea of structural chemistry was popularized in 1860, when the leading chemists of the world attended the first International Chemical Congress at Karlsruhe to resolve their confusions about atoms, molecules, and equivalents. At the close of this 4-day session, Stanislao Cannizzaro clarified the concept of molecules with his reprints on "the message of his old teacher, Avogadro,

* Presented before the Division of Chemical Literature Symposium on Notation Systems, 158th Meeting, ACS, San Francisco, Calif., April 1968.

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who was then dead and forgotten." Alexander Butlerov thereupon predicted that the future task of the chemist was to determine the atomic arrangements in these molecules. He was the first to use the word "structure" in its modern chemical meaning.

THE LINE-FORMULA CONVENTION

Ever since this first recognition of structural importance, chemists have identified the molecular configurations in text copy simply by delineating the symbols for the corresponding molecular segments, one after another as connected. These are the "line-formula" descriptions.

The exact convention relating to "linear expression of formulae" was clearly defined by A. D. Mitchell 20 years ago: "Each full point (or period mark) is regarded as separating two atoms which are directly linked in the main chain of a compound, and atoms or groups attached to each of these atoms are written immediately after it and before the next full point."

Josef Loschmidt led this new era of structural chemistry with his publication of *Chemische Studien* (Vienna, 1861), which contained 368 remarkably astute graphical diagrams, including scores of benzene ring postulations" (four years before Kekulé "introduced" this ring to the chemical world). His text included a few "rationelle Formel" such as $C_2H_5O.C_2H_5$ for ether, $C_2H_5O.CH_3$ for methyl ethyl ether, and $C_2H_5O.C_2H_4O.C_2H_5$ for the diethyl ether of ethylene glycol. In 1863, L. Carius showed a few other comma-separated, line-formula descriptions like $CHHH.O.H$ and $C_2H_5.O.H$,¹⁰ and Emil Erlenmeyer omitted the punctuation with familiar forms like $C_6H_5.OH$, $C_6H_5.I$ and $C_6H_5.CC_2H_5$.¹¹ Meanwhile H. Hübner had used the conventional period punctuation in the first line-formula descriptions to appear in a chemical journal,¹² with his discussion of the $CH_2CN.CO.Br$ and $CH_2Br.CO.CN$ isomers.

In 1866, August Kekulé,¹³ Henry Debus,¹⁴ H. L. Buff,¹⁵ Erlenmeyer,¹⁶ and E. Frankland and B. F. Duppa¹⁷ completed the popularization of this line-formula technique with scores of examples like $CH_3.CO.OH$, $CH_3.CO.CH_3$, $NC-CH_2-CO_2H$, $C_6H_5-CH_2-Br$, and $C_6H_5.SO_2.OH$. Debus described ethylene simply as $CH_2.CH_2$ and acetylene as $CH.CH$; the unsaturation mark that Erlenmeyer introduced at this same time was picked up two years later by C. Graebe¹⁸ in his $C_6H_5-N=N-C_6H_5$ description of azobenzene, and by A. W. Hofmann¹⁹ in his explanation of $-N=C=S$ and $-S-C\equiv N$ isomers. Finally, when H. Wichelhaus introduced the word "valenz" (as a simplification of equivalence or quantivalence) in 1867, he suggested the punctuating refinement of periods to set off "side-group" terminations and hyphens to show directly connected groups, as in his $CO.OH-CH_2.CO.OH$ and $CH_2-CH.OH.CO.OH$.²⁰ Thus, within the short period of just seven years after the birth of structural chemistry in 1861, virtually all of the main ideas relating to line-formula descriptions were conceived and published. No basically new practices appeared for some 79 years. Then, within an identically brief period of just seven years (1947-1954), virtually all of the fundamental features of structure-delineating chemical notations appeared in the international chemical literature.

Examples of these early variations of "rational formulas"

were given with explanatory comments in two other recent reports on chemical structure information processing; hence, they need not be repeated here.^{21, 22}

G. Malcolm Dyson inspired the modern development of chemical notations with his 1947 manual on "A New Notation and Enumeration System for Organic Compounds."²³ An intriguing feature about these first Dyson ciphers was that they were restricted to capital letters, on-line numerals, and 10 punctuation marks. A single letter represented the benzene ring, and five other letters represented similar aromatic skeletons that were used as building blocks for higher elaborations of polycyclic systems. In addition to his manuals, Dyson presented a number of papers on his notation.²⁴⁻²⁶

F. L. Taylor in 1947 developed a mathematically elegant method for enumerating topologically similar positions in all ring systems—carbocyclic or heterocyclic, aromatic or saturated.²⁷ Dyson's greatly enlarged second edition incorporated this Taylor system of ring enumeration after A. M. Patterson pointed out the "Possibilities for a Combined System of Notation and Nomenclature for Organic Compounds."²⁸

The G-K-D ciphers of Gordon, Kendall, and Davison appeared in 1947 and 1948 with their publications on "A New Systemization of Chemical Species"²⁹ and "Chemical Ciphering: A Universal Code as an Aid to Chemical Systematics."³⁰ At that time, all three were employed in the research department of the Dunlop tire firm in Birmingham, England. They evidently designed their system for early versions of computers, for Davison shortly thereafter reported on "Programs and Equipment for Sorting Gordon-Kendall-Davison Punched Cards for Any Structurally Defined Group."³¹ Again in 1957, Davison and Gordon reported on "Sorting of Chemical Groups Using Gordon-Kendall-Davison Ciphers."³² Davison and Dyson's critical views of each other's approaches were aired in the Letters column of *Chemistry & Industry* in 1954.³³

The Gruber Notation was first reported in 1949,³⁴ and for a time attracted the interest of examiners looking for a suitable international chemical notation. Dr. Gruber pleaded with the IUPAC representatives, before and after their Stockholm meeting in 1951, not to be hasty in their selection of an internationally recommended notation. He provided some stimulating suggestions for classification prefixes and prime marks for another notation,³⁵ but did not pursue further development of his own. In February of this year, he responded to our news about this Notation Symposium with his best wishes, and informed us that he will be 82 years old in June!

In 1949, the IUPAC Commission on Codification, Ciphering, and Punched Card Techniques announced its interest in seeking an internationally suitable chemical notation and invited designers to submit their proposals for review in 1951. It enumerated 11 desirable characteristics or "desiderata" for an internationally acceptable chemical notation.³⁶ These were: simplicity of use, ease of printing and typewriting, conciseness, recognizability, ability to generate a unique chemical nomenclature, compatibility with the accepted practices of inorganic chemical nomenclature, uniqueness, generation of an unambiguous and useful enumeration pattern, ease of manipulation by machine methods—for example, by punched cards, exhibi-

tion of associations (descriptiveness) and ability to deal with partial indeterminates.

The six essential qualities that Dyson cited at this same time seem more to the point, in consideration of all the subsequent developments away from chemical nomenclature: First he cited conciseness, then linear expression, uniqueness, adaptability for mechanical manipulation, general simplicity, and recognizability.³⁵

John A. Silk responded to the IUPAC Commission's invitation with his "New System of Organic Notation," distributed privately in 1951. More than a decade later, he published additional details and improvements in his "Linear Notation for Organic Compounds."^{34, 41} Silk had been inspired by Dyson's London lecture of 1946. In 1949, he wrote an article reviewing methods for ring systems, but did not publish it. His interest again returned in 1957 when he served in an ICI company committee to review and comment on the tentative draft of the IUPAC-Dyson notation. He got busy again during the winter of 1960-61 to write the papers published in the *Journal of Chemical Documentation*. J. G. Cockburn, a member of the *British Chemical Abstracts* staff, responded at the same time (1951) with his "Newcastle System." Silk met him around that time and recalls that Cockburn devised his system more as a chemical shorthand than a systematic notation and was not proposing to develop it.

In August 1951, these linear notations of Dyson, Gordon, Kendall and Davison, Gruber, Silk and Cockburn were briefly reviewed by the IUPAC Commission at Massachusetts Institute of Technology, along with a notation that was based on the "Principle of Least Effort."⁴² This notation, first described in 1950, later was reported in *Chemical and Engineering News*,⁴³ in the British ASLIB Proceedings,⁴⁴ and finally was published as a small manual.⁴⁵

F. R. Benson meanwhile had shown how this American notation could be used for "Recording and Recovering Chemical Information with Standard Tabulating Equipment";⁴⁶ and at this same ACS Meeting in 1953, the author described tabulating applications in a toxicity registry.⁴⁷

E. G. Smith started using this notation with standard punched-card equipment in 1952 and in 1954 wrote a faculty report on its use in "A Punched Card Catalog of the Physical Properties of Some Common Organic Chemicals." In 1960, he reported its attractions in "Machine Searching for Chemical Structures."⁴⁸ He was so encouraged with his experiences in encoding some 50,000 structures that he volunteered to start revising the 1954 manual. After seven years of hard labor—laboring mainly with a controlling committee of notation users—he completed a manual that incorporates a number of his own ideas and provides far more comprehensive coverage than the 1954 edition. This new manual⁴⁹ is published by the McGraw-Hill Book Company and is available with additional computer-generated services from the Institute for Scientific Information.⁵⁰

H. T. Bonnett started studying this notation in February 1953, after his interest was aroused by F. R. Benson. Bonnett and D. W. Calhoun in 1961 described the "Application of a Line Formula Notation in an Index of Chemical Structures";⁵¹ and at the same time, Gelberg,

Nelson, Yee, and Metcalf reported on "A Program Retrieval of Organic Structure Information via Punched Cards."⁵² Both groups were collaborating with E. G. Smith, using the notation that the author called to the attention of the ACS Division of Chemical Education as a "New Tool for Teaching Structural Chemistry" at this same national meeting of ACS.⁵³

Alan Gelberg decided to experiment with this "least effort" notation in December 1959 at the Industrial Liaison Office of Edgewood Arsenal. He was instrumental in demonstrating the capability of the notation and the power of its permuted indexing to a number of other subsequent users.

H. Winston Hayward in November 1961 announced a "New Sequential Enumeration and Line Formula Notation System for Organic Compounds."⁵⁴ By this time the IUPAC Commission on Codification, Ciphering, and Punched Card Techniques had adopted the Dyson notation as the basis for a provisional international notation, had published a tentative version in 1958,⁵⁵ and a final version in 1961.⁵⁶ Hayward later reported on some experiences with his notation in a summer training project,⁵⁷ and additional developments are included in this symposium.⁶¹ Dyson also will be reporting on "Modifications and Abbreviations Recommended for Computer and Visual Handling of the IUPAC Notation" in the next paper,⁶² followed by H. F. Dammers and D. J. Polton's "Use of the IUPAC Notation in Computer Processing of Information on Chemical Structures."⁶³

Joshua Lederberg, in Stanford University's School of Medicine in Palo Alto, introduced a "Notational Algorithm for Tree Structures" called DENDRAL-64 (*dendritic algorithm*, 1964), "A System for Computer Construction, Enumeration and Notation of Organic Molecules as Tree Structures and Cyclic Graphs."^{64, 65} Part II, "Topology of Cyclic Graphs," appeared a year later⁶⁶ and his *Systematics of Organic Molecules, Graph Topology and Hamiltonian Circuits, A General Outline of the DENDRAL System* issued in 1966. Unique DENDRAL notations can be generated from the nonunique synonyms through Lederberg's computer programs.

No new and unique chemical notation system has come to our attention since 1964, when the "Survey of Chemical Notation Systems" was published.⁴⁹ Meanwhile Sorter, Granito, Gelberg, and their former associates at Edgewood Arsenal reported on encyclopedic notation indexes made by "permuting" the symbols of our line notation with computer programs.^{67, 68}

F. A. Landee in 1964 also reported on valuable computer applications, such as a "Checker" program that calculates a molecular formula from the WLN and prints out the discrepancies when this does not agree with the input formula.⁶⁹ Errors are about equally divided between notations and formulas! Landee carried these notation ideas to Moscow in 1965, in a paper on "Computer Methods of Handling Files of Chemically Oriented Information."⁷¹ Other impressive notation-programming developments from his Computations Research Laboratory were reported elsewhere⁷² and in this symposium.⁷³

Gibson, Granito, Renard, and Metcalf in 1965 provided a brief introduction to "The Wiswesser Line-Notation," principally for chemistry students and teachers.⁷⁴ A more comprehensive introduction to the various applications

of this WLN was presented at Edgewood Arsenal in October 1966, and these proceedings have just been published. Related reports or joint studies at J. T. Baker Chemical Company and Fort Detrick were published elsewhere in 1966."

J. K. Horner contributed to this Edgewood Arsenal Conference with his separately published report on "Low-Cost Storage and Retrieval of Organic Structures by Permuted Line Notations: Small Collections." The presentations by Ernest Hyde on "A Computer-Generated Open-Ended Fragment Code" and Lucille Thomson on "Structure Display" also were elaborated elsewhere."

This concludes the brief and incomplete historic review of chemical notation developments that were published during the past 107 years. Other more impressive developments reported at this San Francisco meeting, such as those just announced by the Institute for Scientific Information, certainly should assure all observers that chemical notations are here to stay as long as chemists continue to read and write about the substances they are creating and studying.

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